

Magnetospheric Ion Implantation in the Icy Moons of Giant Planets

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NWO visitor (May-July 2012)

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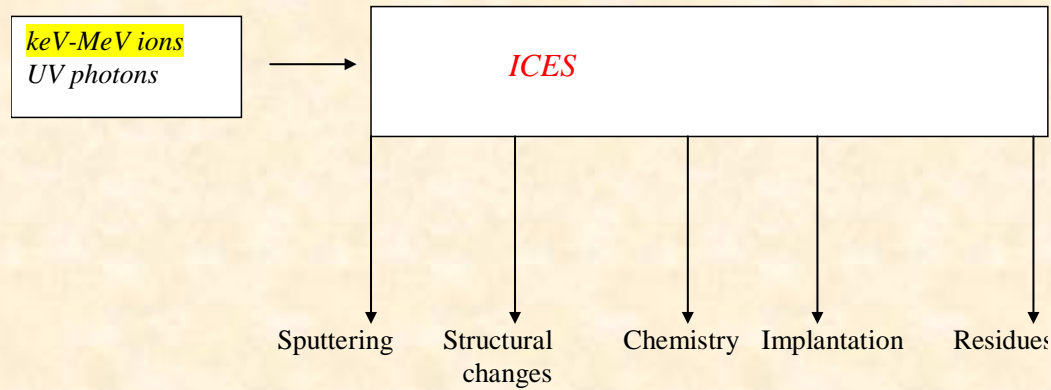
CoI at CIMAP-GANIL, Caen, (France) :

P.Boduch, J.J. Ding, T. Langlinay, X.Y. Lv,

H. Rothard

The experimental work in Catania has been performed by the LASP Team and coworkers (G.A. Baratta, R. Brunetto, D. Fulvio, M. Garozzo, O. Gomis, S. Ioppolo, Z. Kanuchova, G. Leto, P. Modica, L. Moroz, M.E. Palumbo, F. Spinella, G. Strazzulla, et al.).

Experimental study of the effects induced by fast ions on solids of astrophysical interest have been performed in several laboratories in the world



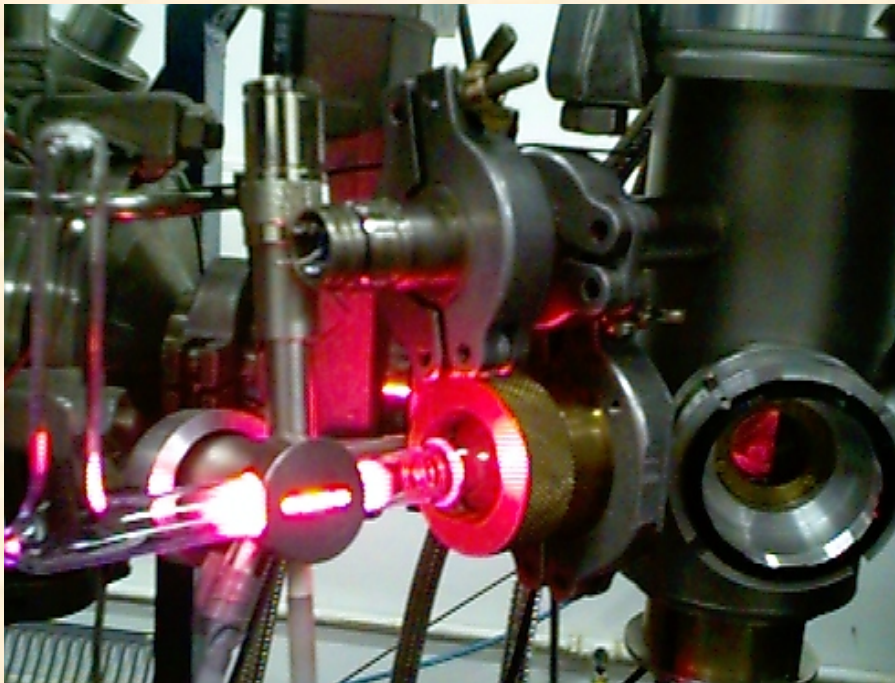
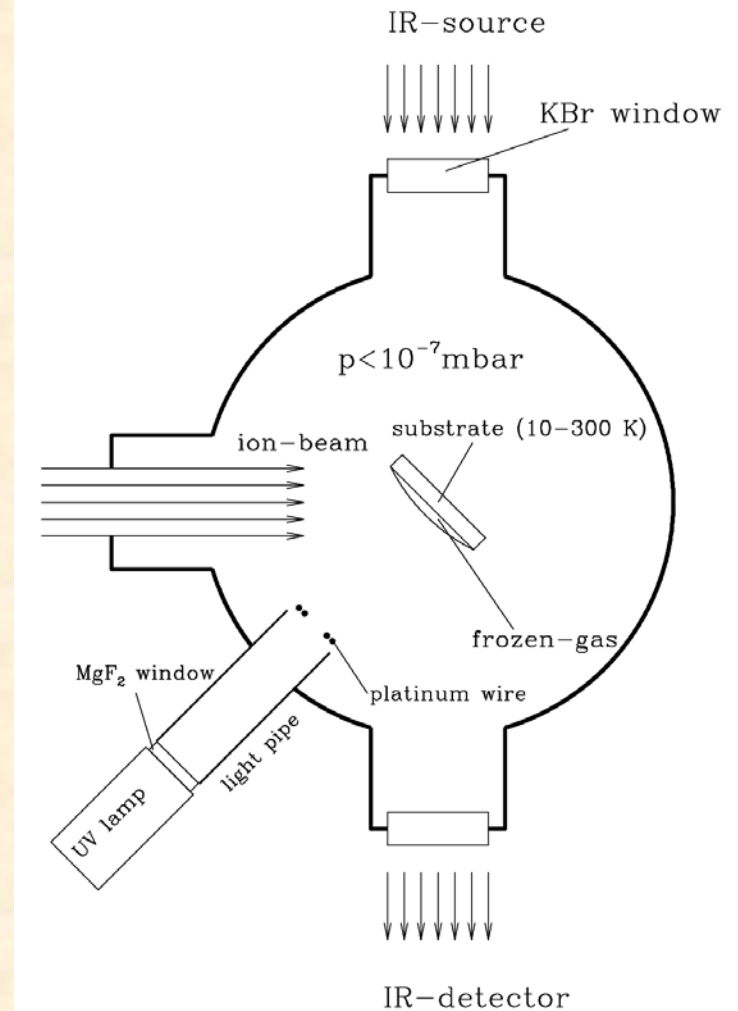
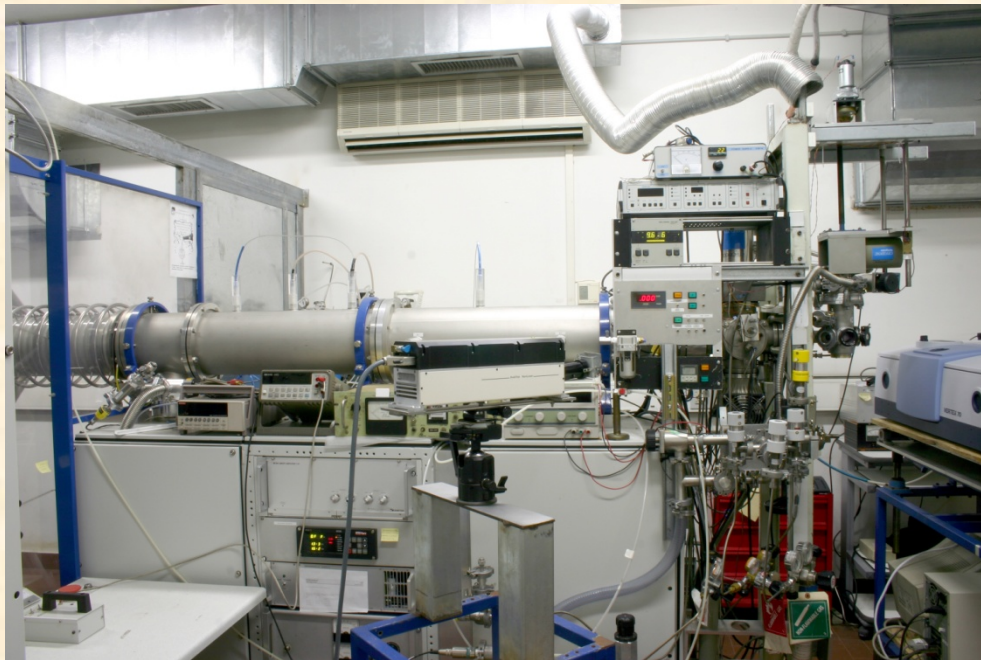
Techniques of analysis :

IR Spectroscopy

Raman Spectroscopy

Mass Spectrometry

In “situ” PROCESSING



Solid Materials in Space

	SILICATES	CARBONS	ICES
<u>WHAT:</u>	olivines piroxenes	PAHs graphite Amorphous carbons	H ₂ O CO ₂ CO NH ₃ CH ₄ OCS
<u>WHERE:</u>	Interstellar medium (ISM) Circumstellar regions Interplanetary medium Planets and satellites Minor objects (asteroids, comets, TNOs)		

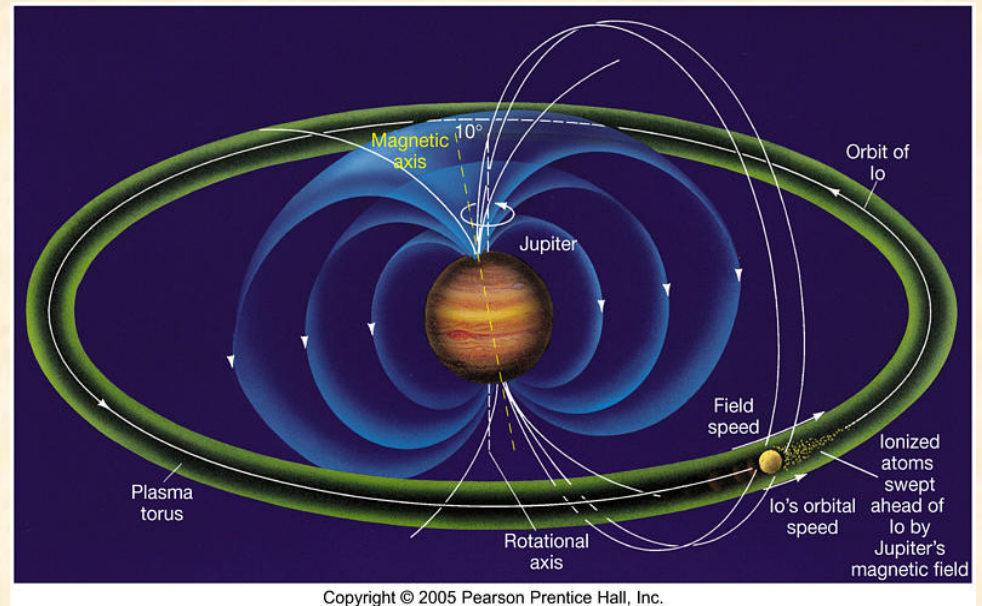
The Jovian magnetosphere

Jupiter's moons are dipped into the magnetosphere and their surfaces are continuously bombarded by energetic ions (mainly H^+ , S^{n+} and O^{n+}) accelerated by Jupiter's magnetic field:

Mean Energy Flux ($keV\ cm^{-2}\ s^{-1}$)

Io	1×10^9
Europa	6×10^{10}
Ganymede	5×10^9
Callisto	2×10^8

Cooper et al., 2001, Icarus 149, 133



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✓ Effects → Sputtering

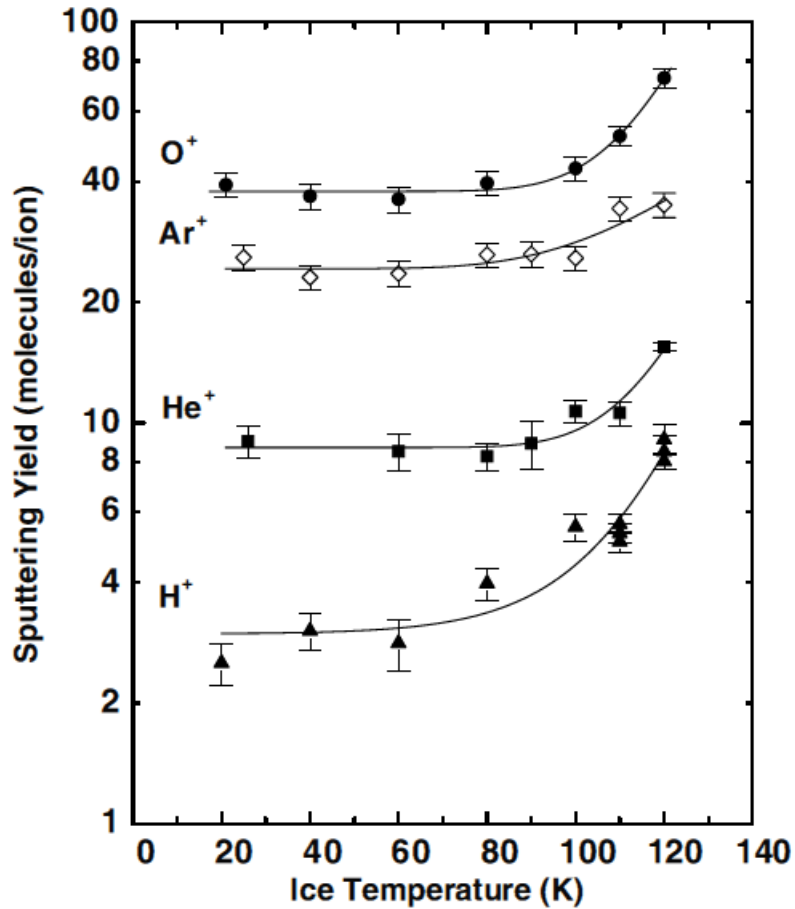


Fig. 2. Sputtering yield of ice for 30 keV ions at normal incidence versus temperature.

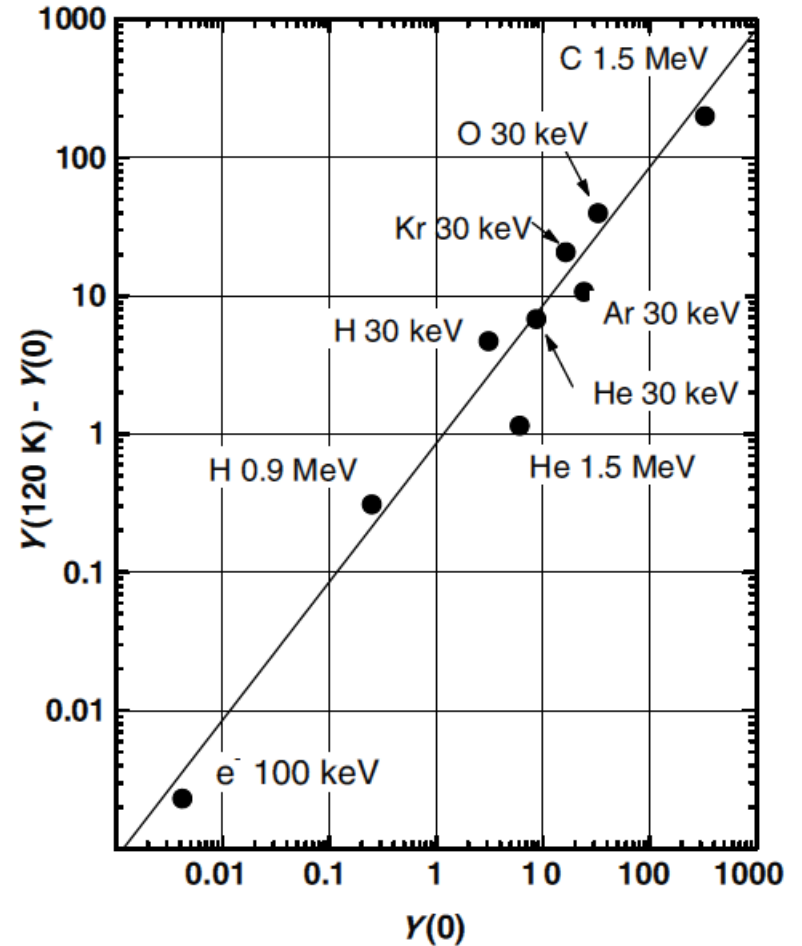


Fig. 3. The difference in the sputtering yield at 120 K and at <90 K: $Y(120) - Y(0)$, versus $Y(0)$. Data for 30 keV H, He, O and Ar ions: this work; 30 keV Kr [21]. Electron data from [45]. MeV data from [8]. The line is a linear fit, $Y(120 \text{ K}) = 0.85Y(0)$.

H₂O ice

Measurements of sputtering yield allow e.g.:

-To calculate the lifetimes for the destruction of icy grains

(e.g. Baragiola RSLTS 362, Issue 1814, 29, 2004)

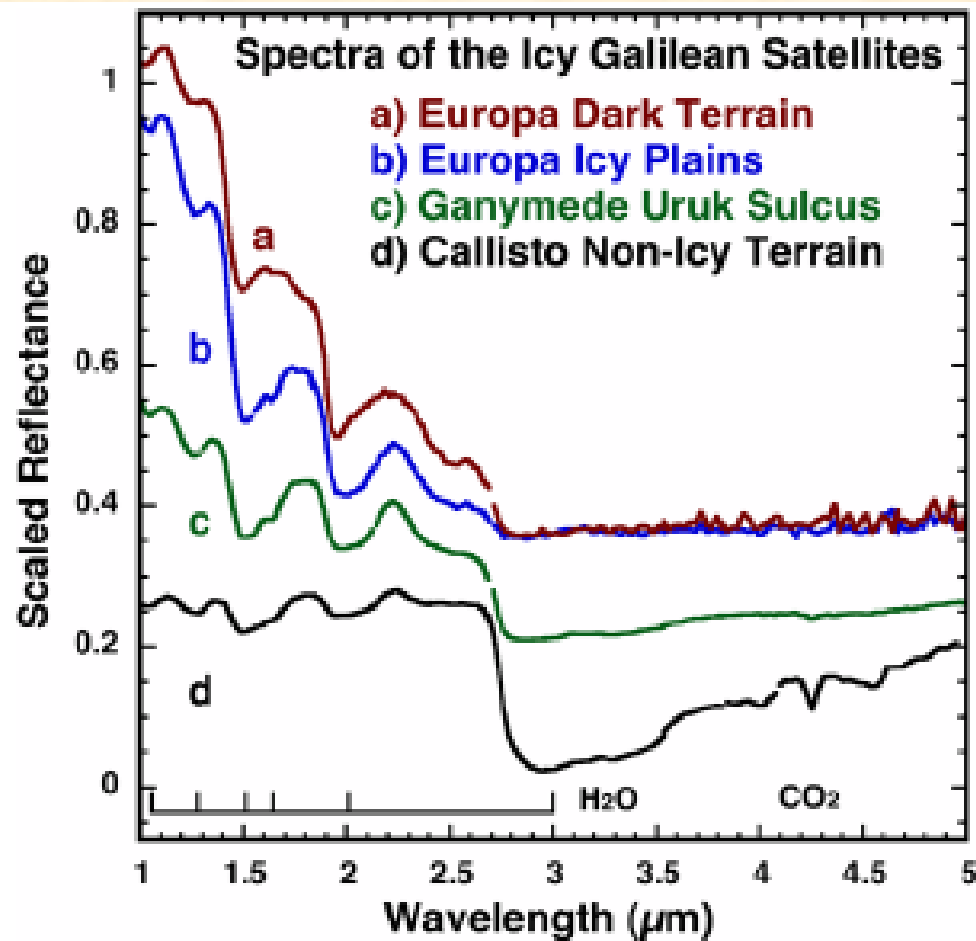
-To predict the column density of O₂ in the atmosphere of Europa

(e.g. Johnson et al., GeoResLett 25, 3257, 1998)

And (hopefully) will allow:

-To explain the presence of ionised and neutral species in the environments of the Jovian Moons **(JUICE!!)**

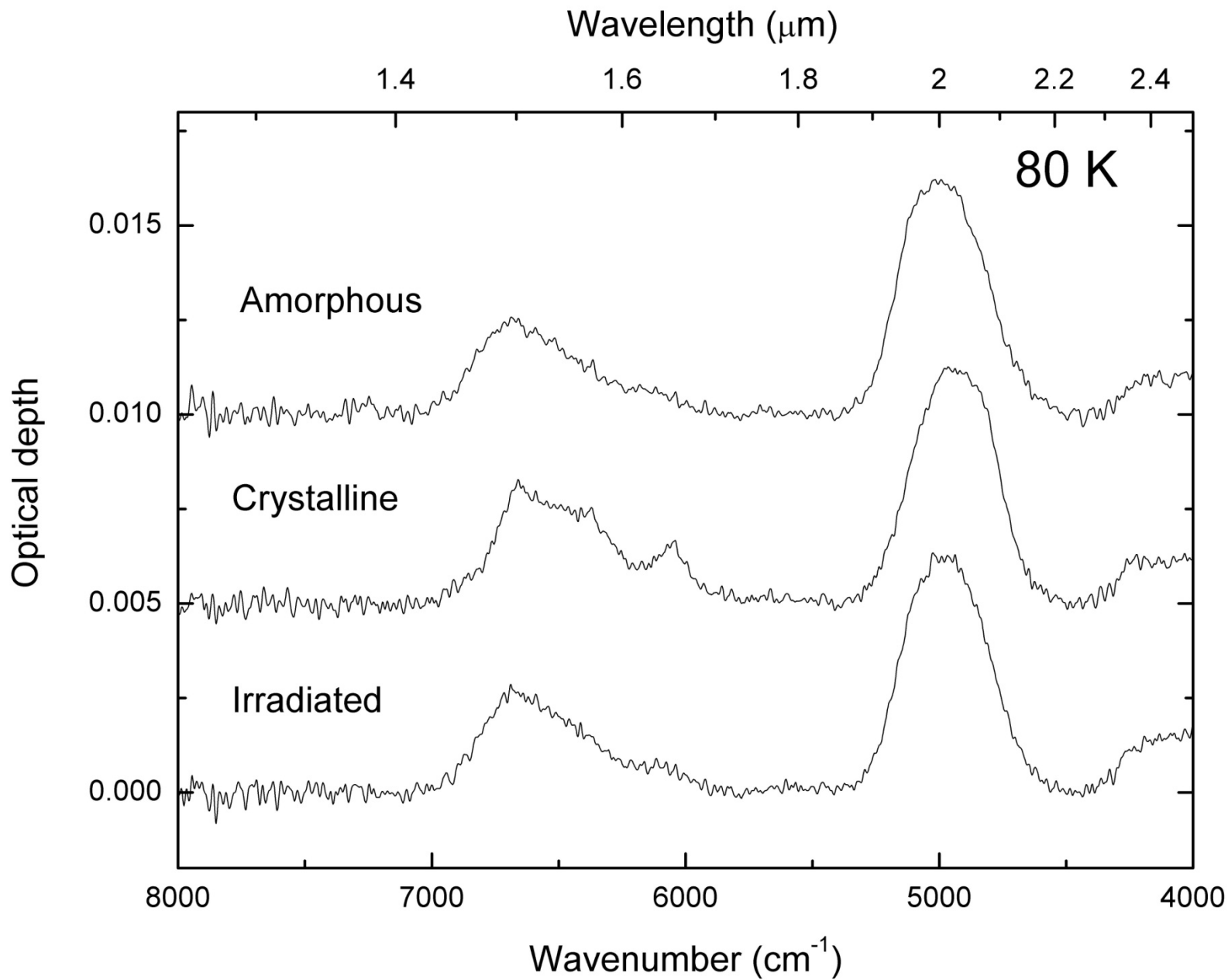
✓ Effects → Structure of the ice



Galileo NIMS :

primary H₂O absorption at ~ 1.0 , ~ 1.25 , ~ 1.5 , ~ 2.0 and ~ 3.0 μm are .

The 1.65- μm crystalline water ice feature is apparent on all three satellites, but is faint in the Europa dark terrain spectra, which also exhibit strongly asymmetric absorption feature shapes at 1.5 and 2.0 μm representative of bound water such as hydrated salts and sulfuric acid hydrate.



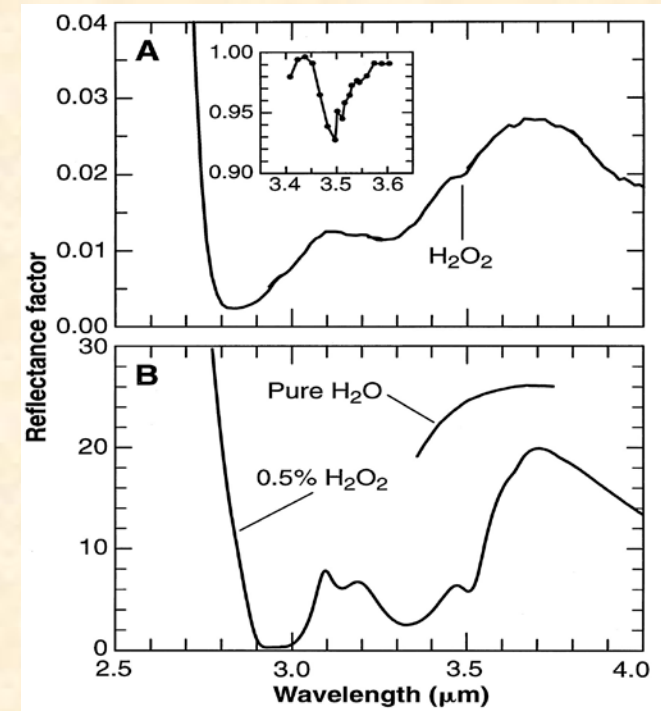
✓ Effects → Chemical

Hydrogen peroxide in the Solar System

Previous Work

Carlson *et al.* (1999) observations

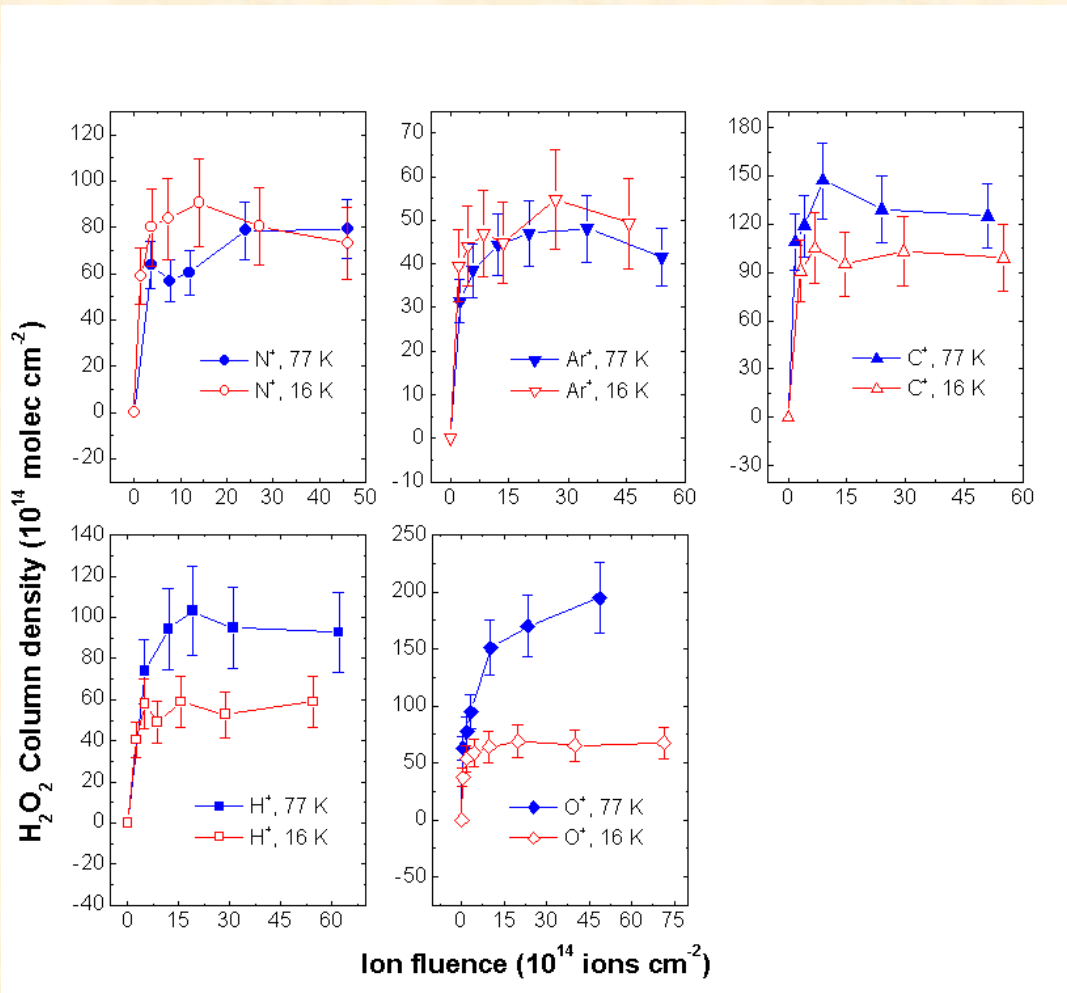
- H_2O_2 found on the surface of Europa by Galileo NIMS spectra
- Radiolysis is supposed to be the dominant formation mechanism



H₂O₂ production as a function of temperature

Ion	Range (Å)
H ⁺	7800
C ⁺	1800
N ⁺	1500
O ⁺	1350
Ar ⁺	700

Ion penetration depth
The ion energy is 30 keV



H₂O₂ production is greater or, at most, equal at 77 K than at 16 K:

- Radical mobility
- Diffusion velocity
- Incoming H and O may form H₂O₂ molecules

O. Gomis, M. A. Satorre, G. Strazzulla, G. Leto

Hydrogen peroxide formation by ion implantation in water ice and its relevance to the Galilean satellites.

PLSpSci 52, 371 (2004)

G. Strazzulla, G. Leto, O. Gomis, M. A. Satorre

Implantation of carbon and nitrogen ions in water ice.

Icarus 164, 163 (2003)

O. Gomis, G. Leto, G. Strazzulla

Hydrogen peroxide production by ion irradiation of thin water ice films.

A & A 420, 405 (2004)

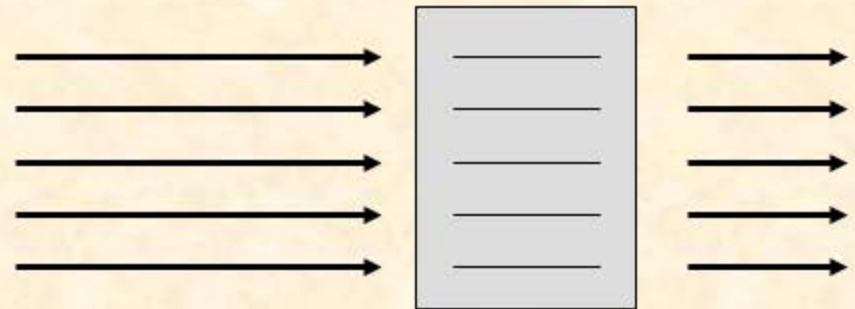
- **Ion irradiation is the primary candidate to explain the presence of H_2O_2 on the Galilean moons**
- **On Europa, H_2O_2 could be concentrated in patches**
 - **The results support the possibility of a radiation-driven ecosystem on Europa**

✓ Effects → Ion Implantation

Two kinds of experiments

❖ Thin film
(beam passes through)

InterStellarMedium

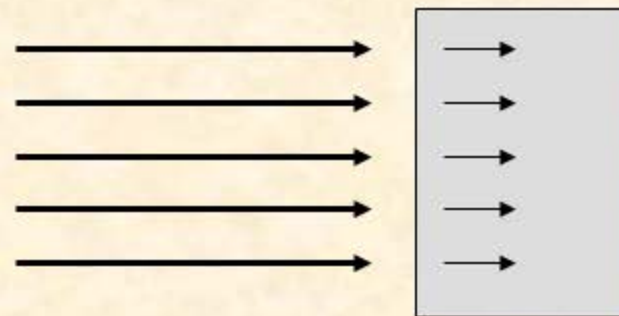


Ions

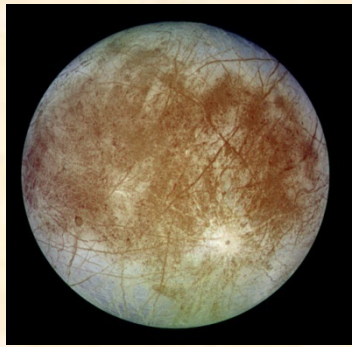
Sample

❖ Thick film
(ion implantation)

Solar System



ION (E in keV)	TARGET T= 16-150 K	MAJOR PRODUCED SPECIES (In bold those containing the projectile)	REFERENCES
H (1.5, 30-100)	CO ₂	CO, H₂CO₃ , CO ₃ , O ₃ O-H in poly-water	Brucato, Palumbo, Strazzulla. Icarus 125, 135, 1997
	SO ₂	SO ₃ , polySO ₃ , O ₃ , elemental S	Garozzo, et al PISpSci 56, 1300, 2008
C ⁿ⁺ (10, 30) n=1-3	H ₂ O	H ₂ O ₂ , CO₂	Strazzulla, et al. Icarus 164, 163, 2003 Lv et al 2012 (submitted)
N (15, 30)	H ₂ O	H ₂ O ₂	Strazzulla, et al. Icarus 164, 163, 2003
	H ₂ O + CH ₄	C₂H₆ , CO, CO ₂ , OCN⁻ , HCN	Strazzulla, PISpSci 47, 1371, 1999
O (30)	CH ₄	C ₂ H ₆ , C ₂ H ₄	Palumbo AdvSpRes 20, 1637, 1997
	N ₂ + CH ₄	C ₂ H ₆ , C ₂ H ₄ , HCN	Ottaviano, Palumbo, Strazzulla, Conf Proc. SIF 68, 149, 2000
S ⁿ⁺ (35-200) n= 1-11	H ₂ O	H ₂ O ₂ H₂SO₄ dissolved in H ₂ O	Strazzulla et al. Icarus, 192, 623, 2007 Ding et al. in preparation



Europa



Ganymede



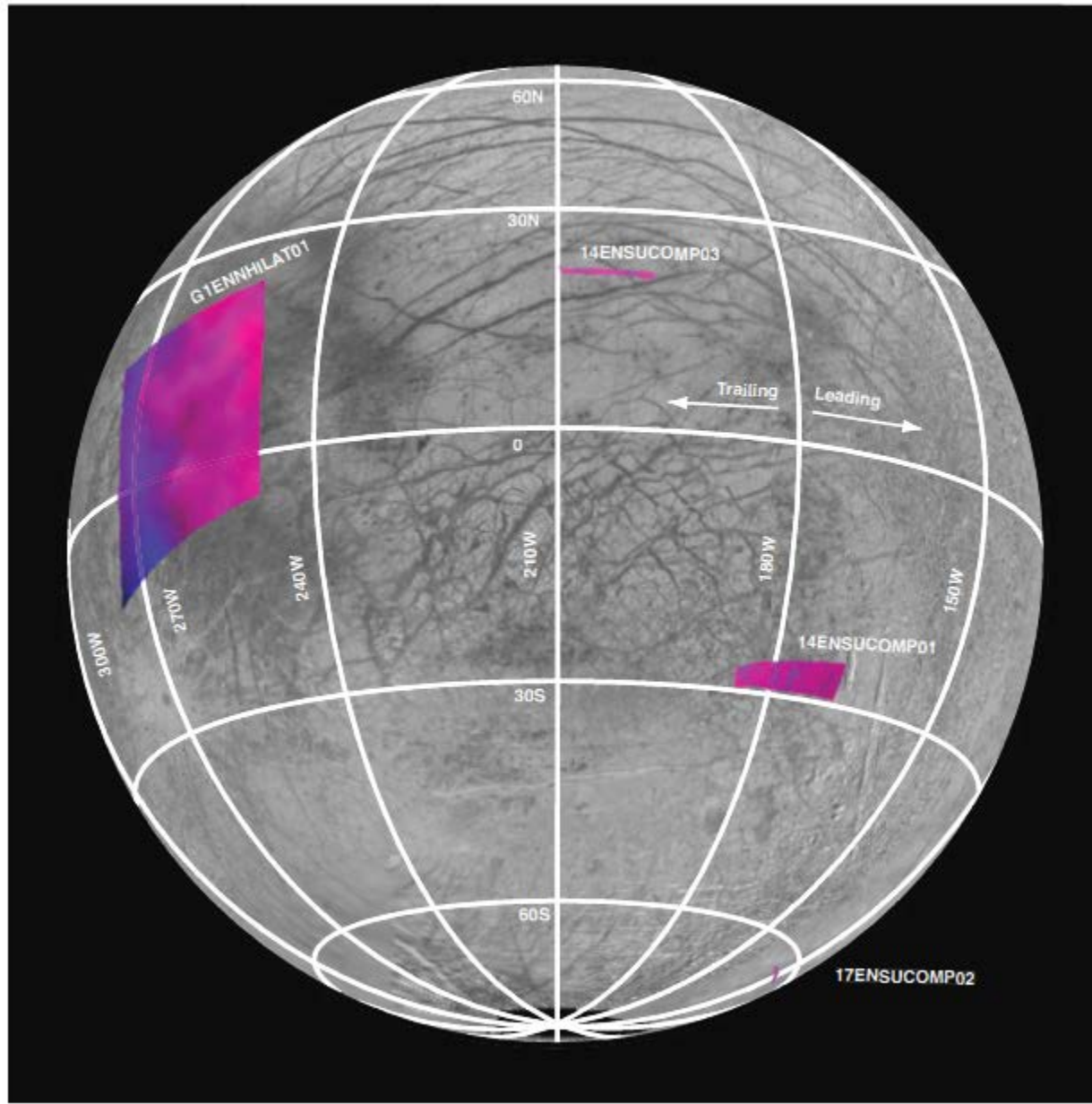
Callisto

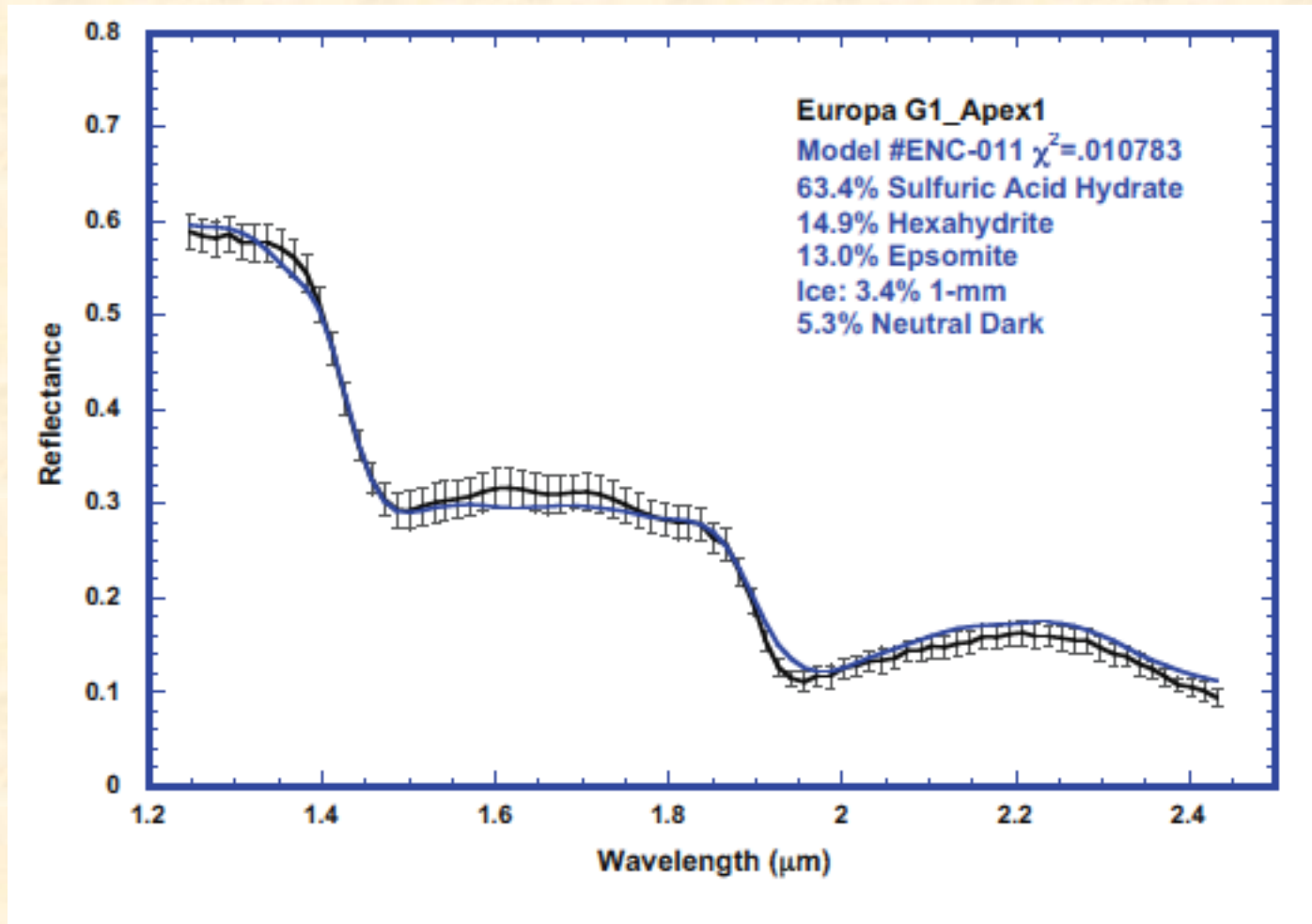
Near IR observations made by the NASA *Galileo* spacecraft showed that:

- On Io SO_2 ice is dominant
- On Europa, Callisto and Ganymede H_2O ice is dominant
- On Europa significant quantities of **magnesium and sodium sulfate and carbonate hydrates**
- Other absorption features and their prime candidates are:
 - **3.4 μm** ($\sim 2940 \text{ cm}^{-1}$) C-H
 - **3.5 “** ($\sim 2857 \text{ cm}^{-1}$) H_2O_2
 - **3.88 “** ($\sim 2580 \text{ cm}^{-1}$) S-H, H_2CO_3
 - **4.05 “** ($\sim 2470 \text{ cm}^{-1}$) SO_2
 - **4.25 “** ($\sim 2350 \text{ cm}^{-1}$) CO_2
 - **4.57 “** ($\sim 2190 \text{ cm}^{-1}$) CN



A still open question is to understand if those species are **native** from the satellites or are **synthesized** by exogenic processes such as ion implantation.



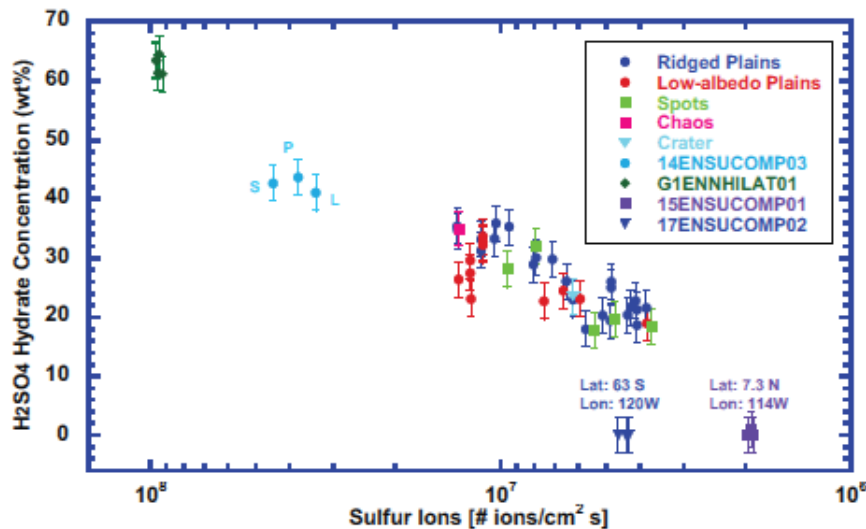
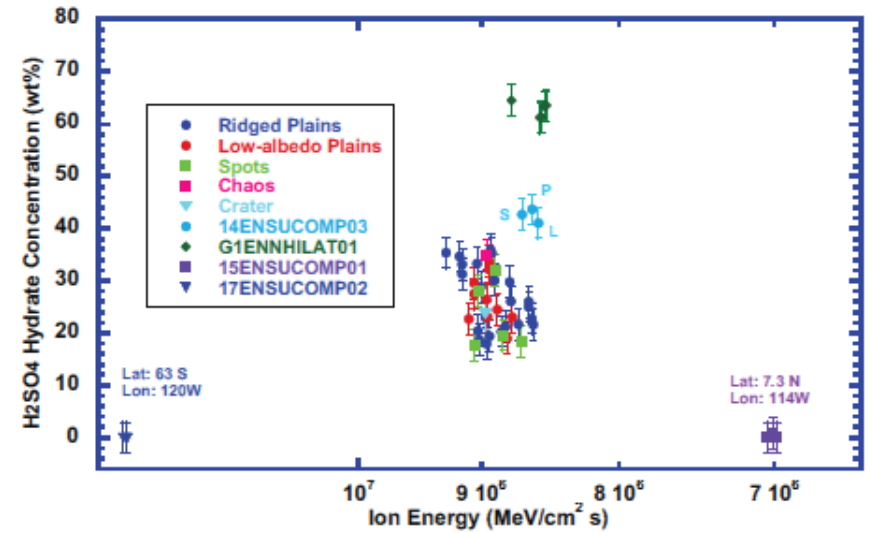
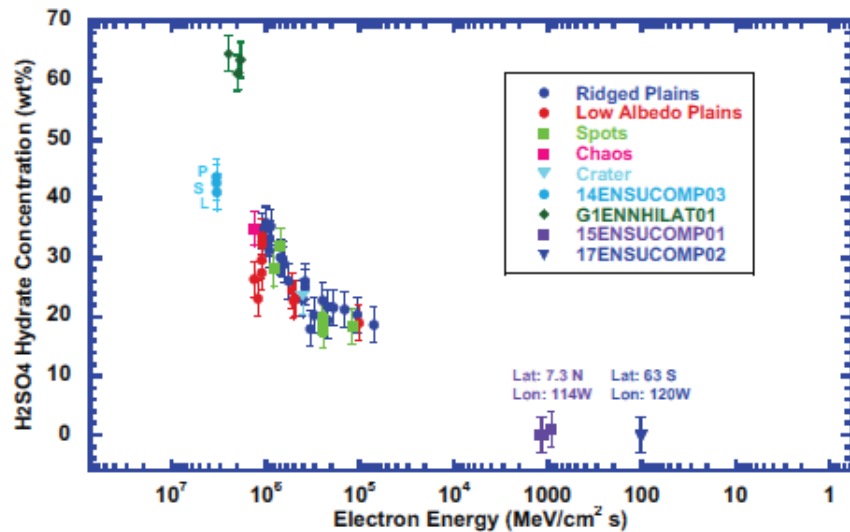


Sulfuric acid (H_2SO_4) Hydrate

Hexahydrite: $\text{MgSO}_4 \cdot 6(\text{H}_2\text{O})$

Epsomite: $\text{MgSO}_4 \cdot 7(\text{H}_2\text{O})$

Dalton et al PSS 2012, in press



Hydrate sulfuric acid after sulfur implantation in water ice

G. Strazzulla ^{a,*}, G.A. Baratta ^a, G. Leto ^a, O. Gomis ^b

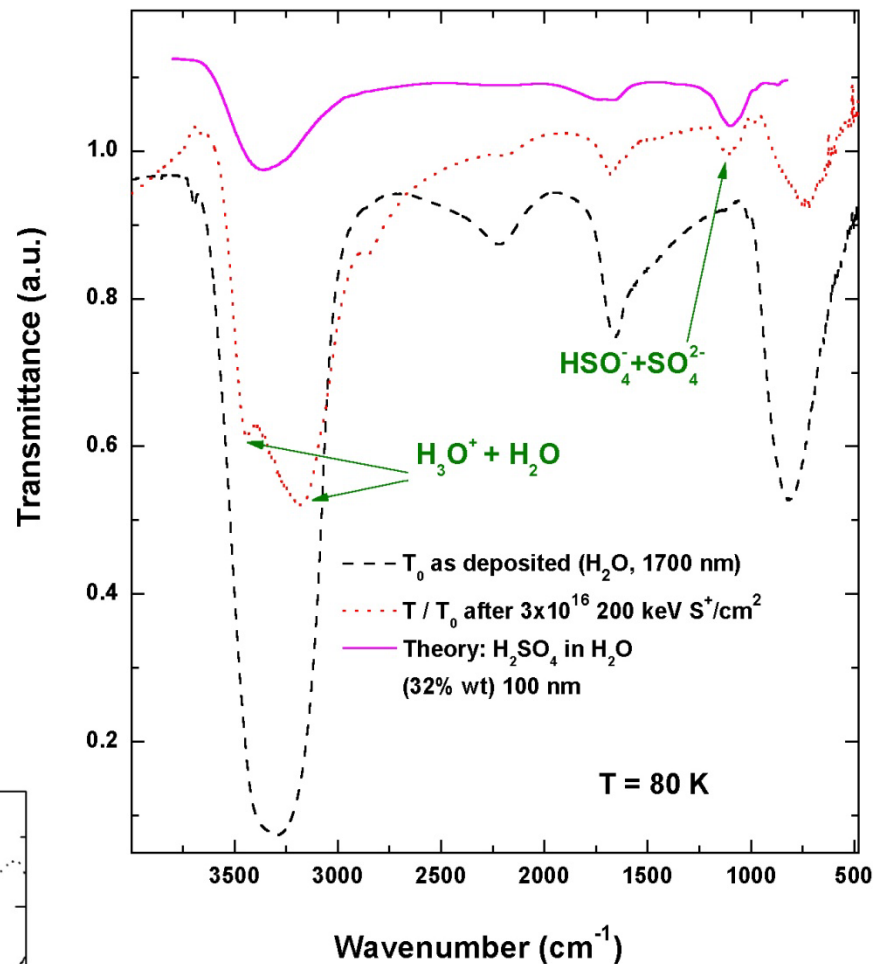
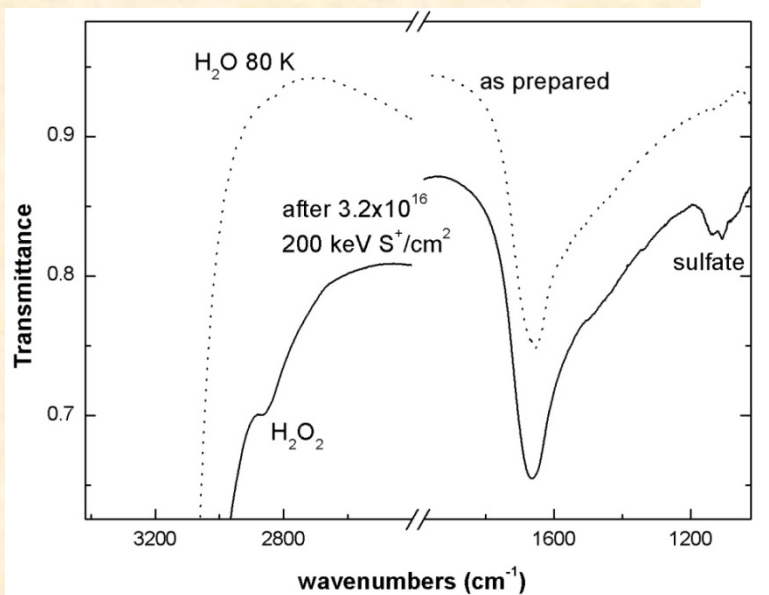
^a INAF – Osservatorio Astrofisico, via S. Sofia 78, I-95123 Catania, Italy

^b Departamento de Física Aplicada, Escuela Politécnica Superior Alcoy, Universidad Politécnica de Valencia, Spain

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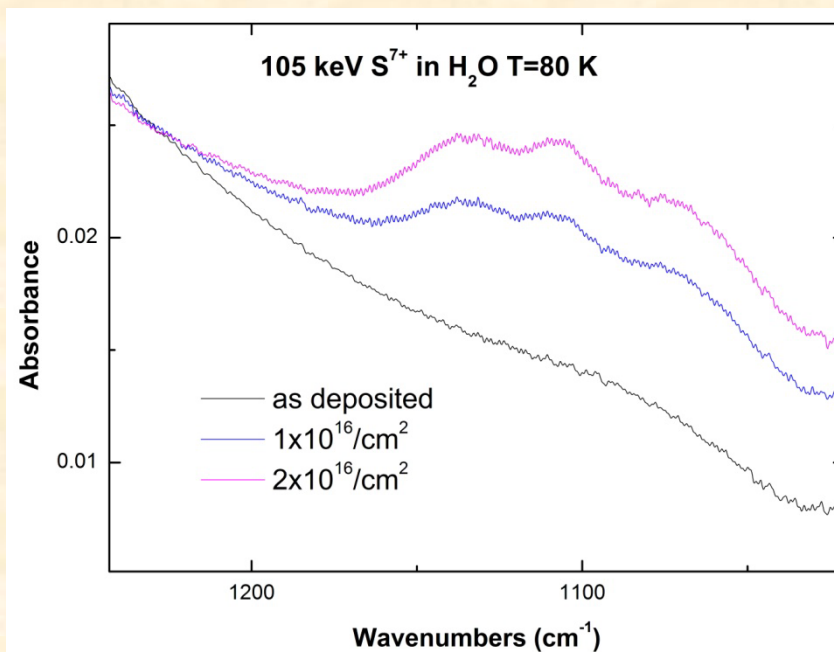
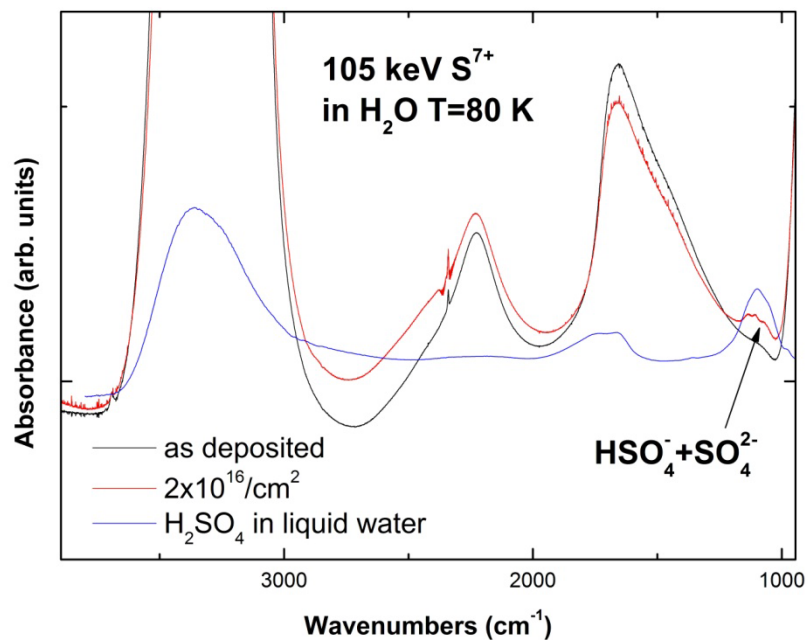
We demonstrate that S-implantation efficiently forms hydrated sulfuric acid whose observed abundance is explained as caused by an exogenic process.



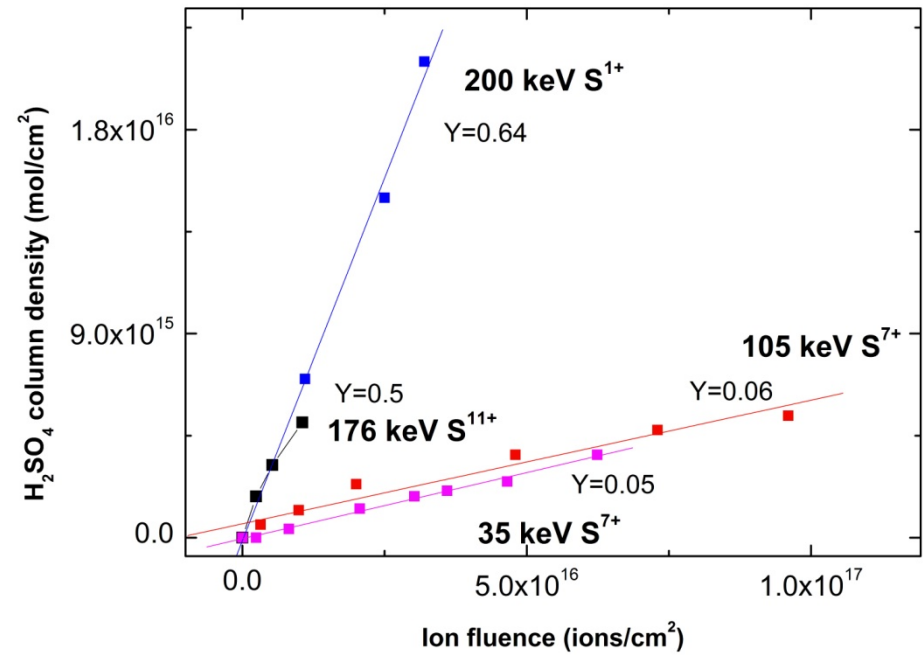
Implantation of S^{n+} ions on the Galilean moons

J.J. Ding (1), P. Boduch (1), A. Domaracka (1), T. Langlinay (1), X.Y. Lv (1),
M.E. Palumbo (2), H. Rothard (1) and G. Strazzulla (2)

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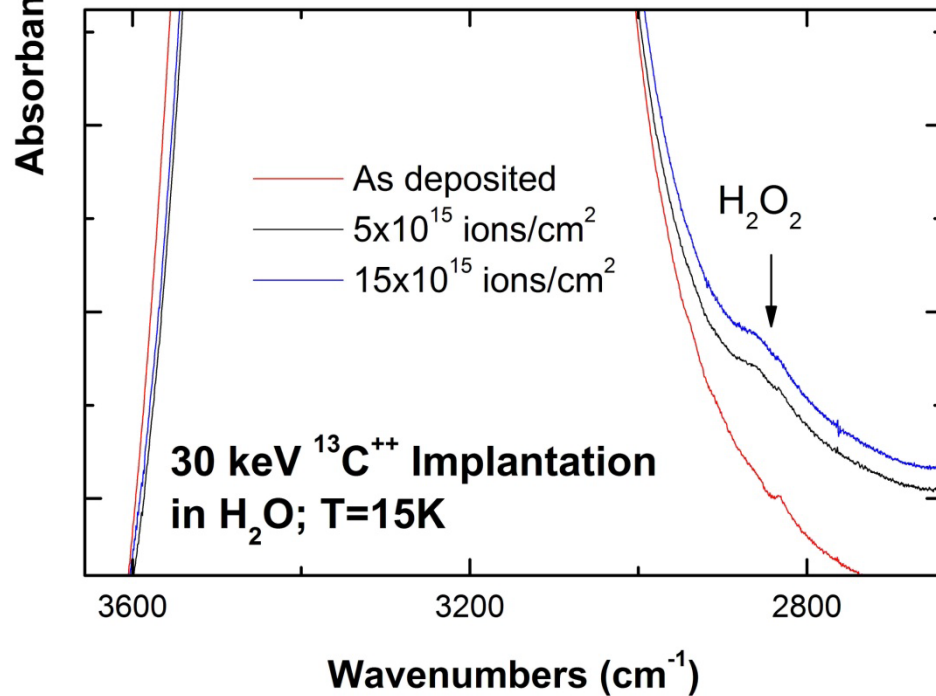
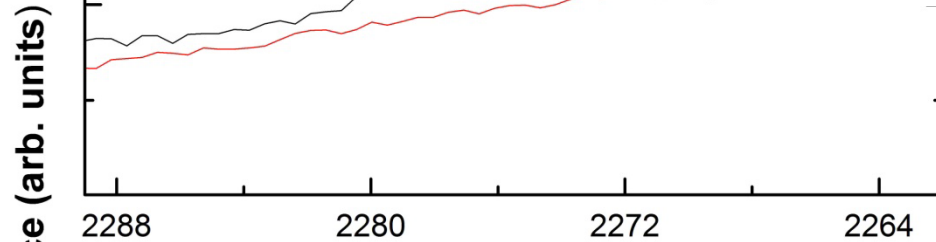
Ding et al. 2012
In preparation



Implantation of C^{n+} ions on the Galilean moons

X.Y. Lv (1), P. Boduch (1), V. Bordalo (2), A. Domaracka (1), T. Langlinay (1),
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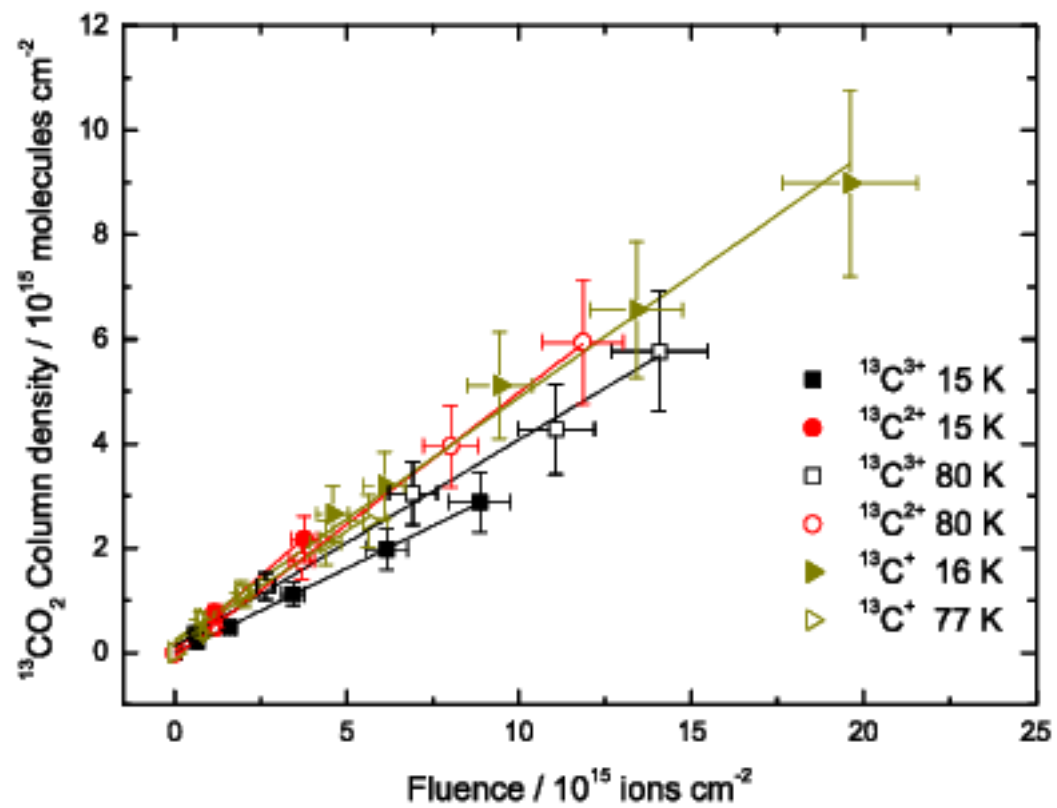


Lv et al. 2012
submitted

Experiments are conducted at
the GANIL Laboratories (Caen,
France)

Available ION Beams: Multi
Charged; E from a few keV to
hundreds MeV

Lv et al. 2012
submitted



Application to icy satellites

- Although a relevant quantity of CO₂ can be formed, this is not the dominant formation mechanism provided that the fluxes of C ions are correct.

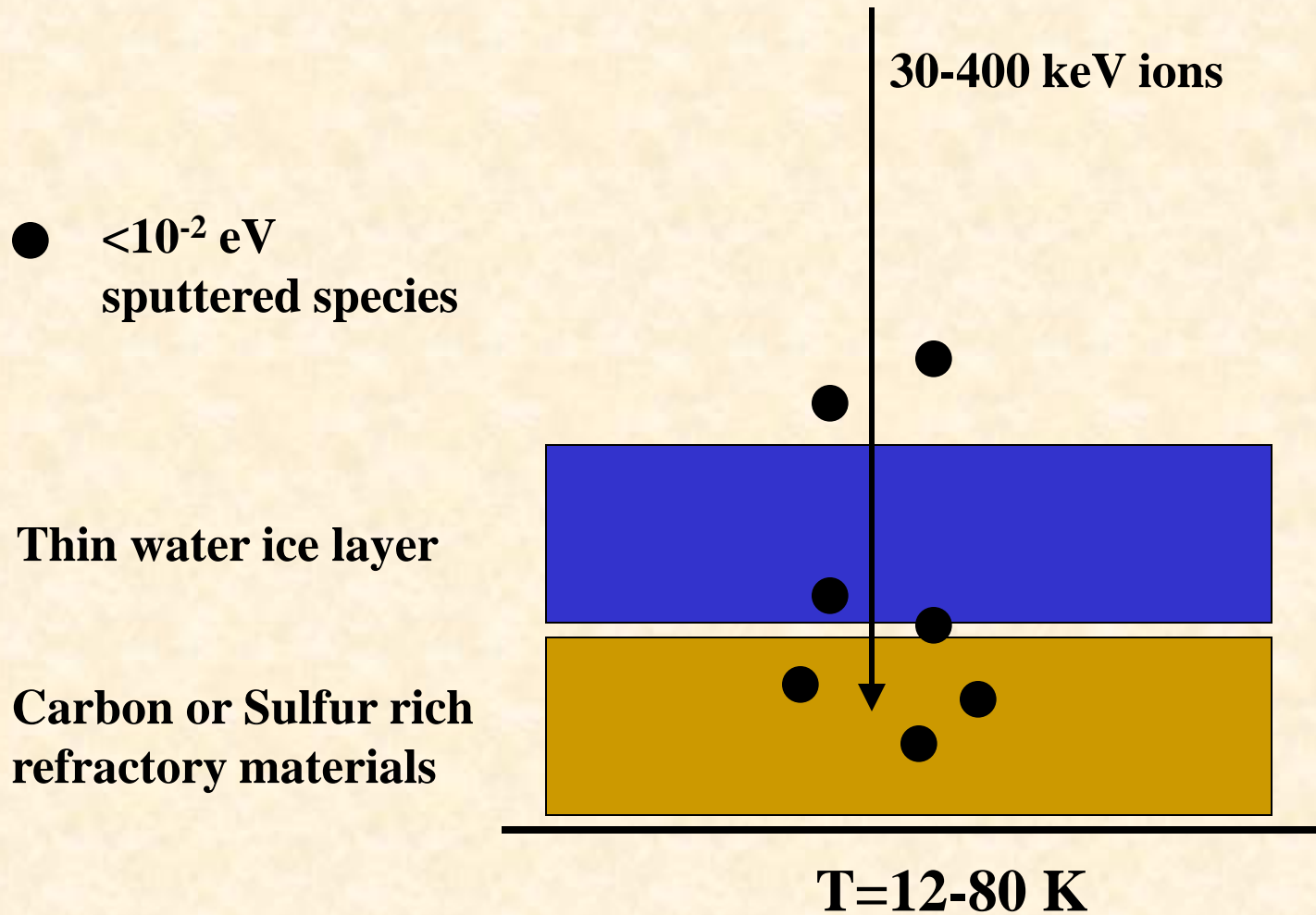
(In particular we would expect more CO₂ at the polar region of Ganymede than at the equator but observations indicate this is not the case)

- Sulfur implantation is the primary mechanism to explain the presence of hydrated sulfuric acid

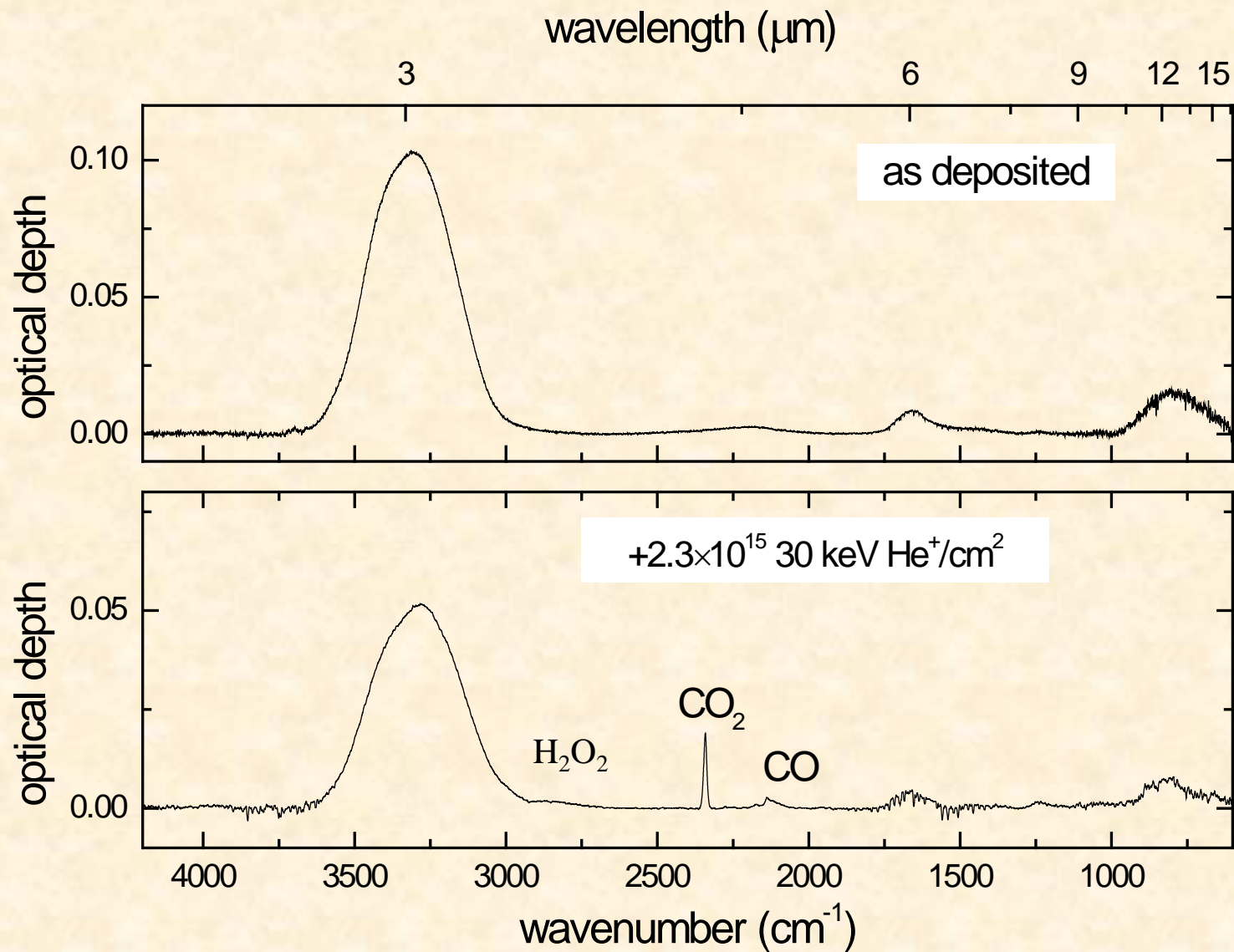
Satellite	S-ions cm ⁻² s ⁻¹	C-ions cm ⁻² s ⁻¹	Time (years) to produce 3x10 ¹⁷ CO ₂ mol cm ⁻²	Time (years) to produce 2x10 ¹⁷ SO ₂ mol cm ⁻²	Time (years) to produce 3x10 ¹⁹ H ₂ SO ₄ mol cm ⁻²
Europa global	0.2-1.4 x10 ⁸	0.4-2.8 x10 ⁷	0.8-5.7 x10 ³	0.2-1.2x10 ⁴	1.0-7.3x10 ⁴
Ganymede polar	3.2x10 ⁶	1.6x10 ⁶	1.5x10 ⁴	1.5x10 ⁴	1.5x10 ⁴
Ganymede equator	5.8x10 ⁴	2.9x10 ⁴	8.2x10 ⁵	8.2x10 ⁵	8.2x10 ⁵
Callisto global	3.6x10 ⁵	3.6x10 ⁵	6.6x10 ⁴	6.6x10 ⁴	6.6x10 ⁴

✓ Effects at the interface solid/ice → mixing

Water ice on C or S-rich substrates



H₂O on amorphous carbon at 12 K



The carbon cycle: initiated by ion processing at the interface s between C-bearing solids and water ice!!

Time necessary to produce the amount of CO₂ molecules of the order of those detected on the surfaces of the Galilean satellites (3×10^{17} molecules cm⁻²) for three different surface depths. The times have been obtained from Fig. 20.5 of Johnson et al. (2004) with the assumptions on the amount of available carbonaceous surface area done in the text.

Satellite	Time (years) to produce 3×10^{17} CO ₂ molecules cm ⁻²		
Europa global	5×10^1	3×10^2	2×10^3
Ganymede polar	3×10^2	4×10^3	4×10^4
Ganymede equator	2×10^4	2×10^5	5×10^6
Callisto global	6×10^3	1×10^5	2×10^6
Surface depth (μm)	10	100	1000

We need Juice!

What next?

Some of the (many) open questions:

- Where SO_2 comes from?
- What produces the UV dark material and its distribution on the surfaces?
- Optical constants of relevant materials
- Inter-relations between sulfur and carbon cycles
-